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Sketch a photometric titration curve for the titration of ...

The absorbance of unknown concentration is, the concentration of known solution is, the variable volume is, the absorbance of known concentration is and the volume of unknown solution is. Substitute for, for, for, for and for in equation (4) Thus, Therefore, the concentration of unknown solution (Quinone) in parts per million in the sample is.

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A 0.4740-g pesticide sample was decomposed by wet ashing ...

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The purpose of dissolving reactants in a solution is often to allow them to come in close contact to react. By dissolving both compounds in water, the ions can collide with one another and react to form an insoluble compound. A 0.1 M HCl solution is dilute acid. A 12 M HCl solution is concentrated acid.

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