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CHE 1401 Lab Manual 13. The solubility of a substance in a solvent at a specified temperature is the maximum weight of that substance that dissolves in a given volume (usually 100 or 1000 mL) of a solvent. It is tabulated in hand-books in terms of grams per 100 mL of solvent; the solvent is usually water.

## **au.ma**

CHE 1402 Lab Manual 2. The units of  $P$ ,  $V$ ,  $T$ , and  $m$  must be expressed in units consistent with the value of  $R$ . The first part of this experiment will be a demonstration of Boyle's law. You will vary the pressure of a fixed amount of trapped air at constant temperature, using a special gas buret.

## **LABORATORY MANUAL FOR GENERAL CHEMISTRY II**

General Chemistry II Laboratory Manual, 2017 Revision 23  
Determination of the Heat of Vaporization of Water Pre-Lab



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Name: Sana Munic Date 1) If air in a sealed flask at  $22^{\circ}\text{C}$  has a pressure of 108 kPa, what is the value of  $k$  using the law of Gay-Lussac? 2) Use the value of  $k$  from question 1 to calculate the pressure of the air in the flask at  $45^{\circ}\text{C}$ .

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Laboratory Manual for General Chemistry I

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Nancy Kerner is Lecturer IV and Coordinator, General Chemistry Lab Program in the Chemistry department. She is Chemistry co-editor of MERLOT where one of her roles has been the establishment of evaluation criteria for online learning objects. She is a co-leader in the MELO 3D project. M.A. Columbia University; Ph.D. Columbia University

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